

Physics 406: Homework 3

1. **Maxwell relations:** We have seen that a small change in the internal energy of a pressure/volume systems is related to the change dS in entropy and dV in volume by $dU = T dS - p dV$. We also saw that a change in the enthalpy H is given by $dH = T dS + V dp$.

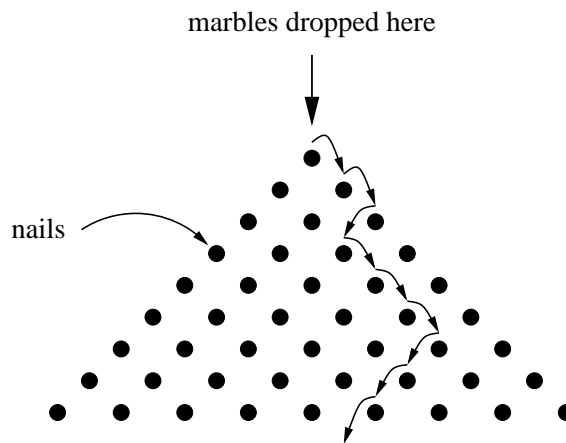
- In week 3 we saw that the free energy F of a system is $F = U - TS$. What is the equivalent expression to the equations above for a small change dF in the free energy?
- By using the fact that the order of partial derivatives doesn't matter, we know that

$$\frac{\partial^2 U}{\partial V \partial S} = \frac{\partial^2 U}{\partial S \partial V}.$$

Use this to prove the **Maxwell relation**

$$\left. \frac{\partial T}{\partial V} \right|_S = - \left. \frac{\partial p}{\partial S} \right|_V.$$

- Use the fact that $\partial^2 F / \partial V \partial T = \partial^2 F / \partial T \partial V$ to prove another similar Maxwell relation.
 - And derive yet a third from the enthalpy. Look at this last relation. Describe briefly what the two sides represent. Remember that a small amount of heat flowing in or out of a system is $dQ = T dS$, and that a process at constant entropy is thus one in which no heat enters or leaves a system.
2. **Combinatorics:** A famous carnival toy is a machine that drops a large number of marbles at the top of a pyramid of nails tacked to a board like this:



The marbles fall down and at each step have a 50% chance of going right or left.

- If there are N rows of nails total, write down an expression for the number of paths $g(N, l, r)$ that a marble can take that make l steps to the left and r steps to the right. Eliminate l and r in favor of the distance $x = r - l$ traveled to the right to get the same number in terms of N and x only.
- If $N = 10$, how many ways are there of traveling distance $x = 10$ to the right? How many ways are there of traveling distance zero?

- (c) If $N = 10$, about how many marbles will have to be dropped before even a single one of them goes all the way to the right-most slot? And how many if $N = 20$?
- (d) When many marbles are dropped, what is the expected mean distance $\langle x \rangle$ traveled, averaged over all of them? And what is the standard deviation of the distance?
- (e) So if you had to say where a single marble dropped would land, between about which values of x would you feel reasonably confident saying it would end up, if $N = 100$? (If you want to be really precise, you could say which values would you have 90% confidence it would land between, but any sensible answer will do for this question. Saying that x lies between -100 and 100 is not a good answer!)

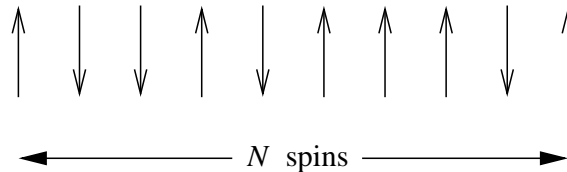
3. **Temperature:** We have seen that for a system of fixed volume and number of particles, the temperature τ and entropy σ are related by

$$\frac{1}{\tau} = \frac{\partial \sigma}{\partial U},$$

where U is the internal energy. Thus we can calculate a rough value for the temperature of a system by comparing a small change ΔU in the internal energy with the corresponding change $\Delta \sigma$ in the entropy thus:

$$\tau = \frac{\Delta U}{\Delta \sigma}. \quad (1)$$

- (a) For the system of magnetic “spins” (i.e., dipoles) that appears in Chapter 1 of Kittel and Kroemer’s book:



write down the internal energy U of the system in a magnetic field B in terms of the magnetic moment m of one of the dipoles, and the spin excess s defined by

$$2s = N_{\uparrow} - N_{\downarrow},$$

where N_{\uparrow} and N_{\downarrow} are the numbers of spins pointing up and down respectively. Also write down the multiplicity $g(N, N_{\uparrow})$ in terms of s . What is the entropy in terms of s ?

- (b) What is the change in energy when the spin excess goes from s to $s + 1$? Show that the corresponding change in entropy is

$$\Delta \sigma = \ln \frac{\frac{1}{2}N - s}{\frac{1}{2}N + s + 1}.$$

- (c) Calculate a rough expression for the temperature τ of the system from Eq. (1) above. What is the value of τ if $N = 20$, $s = 2$, and $m = 1$, $B = 1$?